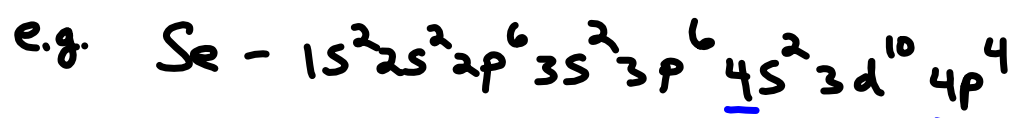


* Electron Dot Diagrams



73.



Rate of Reaction: Definition

Examples - speed $\frac{100 \text{ km}}{\text{h}} \rightarrow \frac{\Delta \text{ distance}}{\Delta \text{ time}}$

lab - $\frac{\Delta \text{ size of balloon}}{\Delta \text{ time}}$

Chemical Reactions - $\frac{\Delta \text{ mass of product/reactant}}{\Delta \text{ time}}$

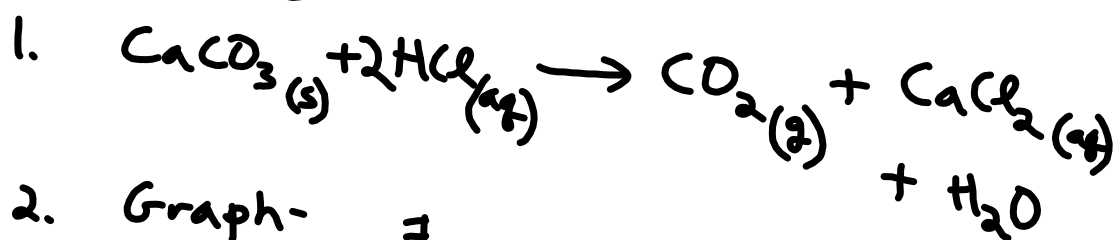
Colour, temperature
Voltage, volume

Read text
p. 462-463
Read rate lab

Rate Laboratory Activity

Data -

Time (30 sec.)	Circumf.
0	2.0
1	
2	
3	



2. Graph -

