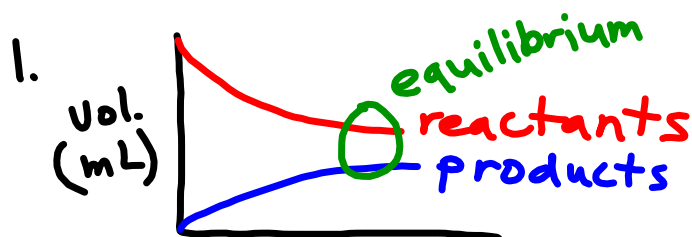


Equilibrium Analogy Questions



Transfers

1. Rate of "converting" gets slower.
2. Rate of converting products to reactants also slows.
3. At the beginning, products are favoured.
4. Curves are flat at equilibrium.
5. Rate of collisions for the forward reaction stays the same so the number of collisions is also same for reverse.
6. The amounts of reactants + products stays the same. Because of the reverse "reaction" the reactants are never used up.

Read text p. 488 - 490, 492, 493